## Pentacyclo[5,3,1,0<sup>2,6</sup>,0<sup>3,5</sup>,0<sup>4,9</sup>]undecane and Tetracyclo[5,3,1,0<sup>2,6</sup>,0<sup>4,9</sup>]undecane

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Summary Direct irradiation of tricyclo [5,5,1,04,9] undeca-2,5-diene in ether afforded pentacyclo [5,3,1,02,6,03,5,04,9] undecane which was hydrogenated to give tetracyclo-[5,3,1,02,6,04,9] undecane.

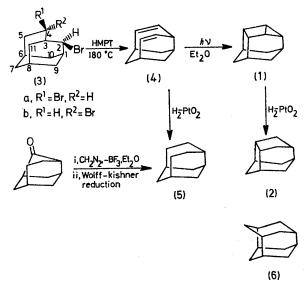
In connection with the chemistry of strained  $\sigma$ -bonds, bicyclo[2,1,0]pentane and its derivatives have attracted considerable attention in the past decade.<sup>1</sup> We now report the synthesis of pentacyclo[5,3,1,0<sup>2,6</sup>,0<sup>3,5</sup>,0<sup>4,9</sup>]unde-

cane (1) which includes the bicyclopentane unit in its framework as well as the tetracyclo  $[5,3,1,0^2,6,0^4,9]$  undecane (noriceane) unit (2).

When the stereoisomeric 2,4-dibromohomoadamantanes (3a) and/or (3b)<sup>2</sup> were treated with hexamethylphosphoric triamide (HMPT)<sup>3</sup> at 180 °C for 3 h, dehydrobromination accompaning skeletal rearrangement took place easily to give the tricyclic diene, (4), m.p. 155—158 °C,† in 85—90% yield: m/e 146 ( $M^+$ , 84%) and 80 (100%); <sup>1</sup>H n.m.r.:  $\delta$ 

† All m.ps. were measured in sealed capillary tubes.

(CCl<sub>4</sub>) 0.80-3.13 (10H, br complex m) and 5.50-6.17 (4H, m);  $^{13}$ C n.m.r.:  $\delta$  (CDCl<sub>3</sub>) 137.0 (2CH=), 132.8 (2CH=), 39.4 (2CH<sub>2</sub>), 38.2 (CH), 32.2 (2CH), 26.9 (CH), and 24.0  $(CH_2).$ 



The skeletal structure of (4) was further supported by the fact that hydrogenation of (4) over PtO2 gave the saturated hydrocarbon (5)  $[m/e \ 150 \ (M^+, \ 100\%); \ ^1H \ n.m.r.: \delta \ (CCl_4)$ 0.90—3.10 (18H, br complex m);  ${}^{13}$ C n.m.r.:  $\delta$  (CDCl<sub>3</sub>) 41.4 (CH<sub>2</sub>), 31·3 (2CH<sub>2</sub>), 30·1 (2CH<sub>2</sub>), 28·1 (2CH<sub>2</sub>), 26·6 (2CH), 26.5 (CCH), and 25.3 (CH) which was identical (H n.m.r.) with authentic tricyclo[5,3,1,04,9] undecane, m.p. 242-246 °C, which was prepared from 2-protoadamantanone by ring

enlargement using CH<sub>2</sub>N<sub>2</sub> (catalysed by BF<sub>3</sub>-EtO<sub>2</sub>)<sup>4</sup> followed by Wolff-Kishner reduction of the resulting ketone.‡

Direct irradiation of (4) in ether with quartz-filtered light for 18 h induced an intramolecular [2+2]cycloaddition<sup>5</sup> to afford the saturated compound (1), m.p. 183-186 °C, as the sole product in 85% yield. Compound (1) in CCl<sub>4</sub> showed an i.r. band at 3100 cm<sup>-1</sup> indicative of a cyclopropane ring. Other spectral data are also compatible with structure (1)  $[m/e \ 146 \ (M^+, 52\%) \ and \ 80 \ (100\%); \ ^1H \ n.m.r.: \delta \ (CCl_4)$ 0.50-2.90 (14H, complex m);  $^{13}$ C n.m.r.:  $\delta$  (CDCl<sub>3</sub>) 43.3 (CH<sub>2</sub>), 38·3 (2CH), 33·1 (2CH<sub>2</sub>), 30·3 (2CH), 22·5 (CH), 20·0 (CH), and 7.3 (2CH)].

When (1) was hydrogenated over PtO<sub>2</sub> in MeOH, 1 mol. equiv. of hydrogen was absorbed to give compound (2),§ m.p. 212—215 °C  $[m/e 148 (M^+, 100\%); {}^{1}H \text{ n.m.r.}: \delta (CCl_4)$ 0.60—2.67 (16H, br complex m);  ${}^{13}$ C n.m.r.:  $\delta$  (CDCl<sub>3</sub>) 41.9 (2CH), 40.9 (CH<sub>2</sub>), 38.1 (2CH), 33.7 (CH), 30.9 (2CH<sub>2</sub>), 30.5 (CH), and 29.5 (2CH<sub>2</sub>)]. The disappearance of the i.r. band at 3100 cm<sup>-1</sup> also indicated cleavage of the cyclopropane ring. This hydrogenation is in good agreement with the well known fact that cleavage of the central bond of bicyclo-[2,1,0]pentane ring system occurs easily in a number of reactions.6

The <sup>13</sup>C n.m.r. data are conclusive in determining the structures of (1), (2), (4), and (5), all of which have a mirror plane.

The fixed bicyclopentane ring in the cage structure is interesting in that any attacking species must approach the bicyclopentane unit from the exo side alone and, in addition, isomerization to a cyclopentene derivative is impeded on the stereochemical grounds (e.g. Bredt's rule<sup>7</sup>).

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† This ketone gave satisfactory elemental analysis and consistent i.r., 1H n.m.r., and mass spectra. However, the position of carbonyl function has not been determined yet.

Compound (2) could be named noriceane. For iceane (6), see C. A. Cupas and L. Hodakowski, J. Amer. Chem. Soc., 1974, 96, 4668; D. P. G. Hamon and G. F. Taylor, Tetrahedron Letters, 1975, 155.

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